# Studies on the Interaction of Natural Antifungals with Metal Ferrocyanides and Their Medicinal Applications

Abstract: Manganese, Silver and Titanium ferrocyanides were synthesized and characterized by elemental and spectral studies. The stabilities of these metal ferrocyanides were investigated in the presence of acids, bases, organic solvents, tap and sea water at room and boiling temperature. The natural antifungal plants studied were azadirachta indica (Neem), ocimum sanctum (tulsi), cassia obtusifolia (money bush), cassia alata (canicro bush), tagetes patula (marigold). The natural antifungal plant extract with metal ferrocyanides complexes were found to be have more antifungal property in comparison to metal ferrocyanides and natural antifungals alone. Antifungal activity of natural antifungals, metal hexacyanoferrate(II) compound and natural antifungal metal ferrocyanide complexes were tested by well known cultured fungus (Aspergillus Niger). The titanium ferrocyanide with neem extract and manganese ferrocyanide with money bush extract complexes were found to have maximum and minimum antifungal property, respectively.

[Nature and Science. 2009;7(3):1-7]. (ISSN: 1545-0740).

Keywords: metal ferrocyanides, natural antifungals, aspergillus niger, medical value, skin infection.

#### 1. INTRODUCTION

Antifungal extracts are to be obtained from various local plants in Guyana. These plants include azadirachta indica, cassia alata, cassia obtusifolia, oscimum sanctum and togetes patula. Studies on these plants have shown that they possess antifungal compounds mainly in their leaves, bark and fruits. The extract may show various activities depending on the method of extraction - such methods include wet, dry and steam distillation. In addition they are also used for various medicinal purposes such as laxative and antibacterial among others. Azadirachta indica is believed to be an answer to many incurable diseases. A mixture of the extracts from the leaves, bark, fruit and seeds is used efficiently to treat skin diseases1. Cassia alata leaves and flowers are used as remedy for asthma, bronchitis, diabetes, ulcers, scabies and skin diseases such as puritis eczema, etc.<sup>2,3</sup>. Cassia obtusifolia leaves extracts are known to be used to treat feet rashes, lotta, scabies, ringworm, and other skin infections. The oil extract of Ocimum sanctum consists of eugenol, eugenol methyl ether and carvacrol which are main contributors to the medicinal value of Tulsi. It helps to eradicate ringworm and other skin diseases when applied to such skin infections4. The whole herb of tagetes patula is used in coughs and dysentery, taken internally in the form of a decotion. Extracts of Marigold can be used as fungicides. In addition to antifungal properties plants extract also have various other medicinal uses<sup>5-10</sup>. To analyze the nature of metal ferrocyanides and the extracts, the best suited fungal spore is Aspergillus niger. A. niger is omnivorous and one of the most common easily identifiable species of the genus. A. niger may also be a common laboratory contaminant 11.

Primitive earth atmosphere was anoxygenic and reducing potential of atmosphere was not high enough hence metals like iron, chromium, molybdenum, manganese and tin etc. were in the form of their lower oxidation states. Considering the fact that cyanide was formed in all simulated Commented [HK1]: Check throughout the document - Scientific names are always written in italics. The generic name is always capitalized, while the species epithet is never capitalized. e.g. Azadirachta indica.

experiments of primitive earth conditions, cyanide could have combined with a large number of metal ions present in primeval sea. Consequently, several insoluble metal ferrocyanides of general formula  $M_2[Fe(CN)_6] \cdot x \; H_2O$ , where  $M=Fe,\, Cr,\, Mo,\, Zn,\, etc.$  could have been formed. It is well established that metal ferrocyanides acts as adsorbents  $^{12}$ , ion-exchangers  $^{13}$  and photosensitizers  $^{14}$ .

Literature survey indicates that no report is available on medical value of natural antifungal – metal ferrocyanide complex. In view of this attempt were made to study medical application of these complexes. In addition present work describes synthesis, characterization and medical application of manganese, silver, titanium ferrocyanides – natural antifungal complexes.

## 2. EXPERIMENTAL SECTION

#### 2.1 Chemicals

All chemicals used were of AnalaR grade and used as such without any further purification. Potassium ferrocyanide, manganese chloride, silver nitrate, titanium tetrachloride were obtained from BDH, Poole, England. Solutions were prepared in doubly distilled water.

#### 2.2 Synthesis of metal ferrocyanides

Manganese and silver ferrocyanides were prepared by Kourim's method<sup>15</sup>. Whereas titanium ferrocyanide was prepared according to the procedure reported by Bastian et al. <sup>16</sup>.

The manganese and silver ferrocyanides were prepared by adding potassium ferrocyanide (167 ml; 0.1 M) slowly to metal chloride/nitrate (500 ml; 0.1 M) with constant stirring. Reaction mixture was heated on water bath for 2-3 h and cured for 24 h. The precipitate was washed with distilled water and dried at  $\frac{60 \text{ C}}{100}$ . The dried product was ground and sieved to  $\frac{125 \text{ } \mu \text{m}}{100}$  BSS mesh size. In case of silver ferrocyanide all reactions were performed in the dark. Silver ferrocyanide was kept in the dark bottle.

The best condition for the preparation of titanium ferrocyanide involves variation in the mole ratio of titanium to hexacyanoferrate(II), which vary between 10 to 1 and 1 to 10, respectively. For this experiment we will use a 0.5 M solution of titanium tetrachloride in 2.0 M aqueous hydrochloric acid and 0.34 M solution of hexacyanoiron(II) acid. The solution of hexacyanoiron(II) acid is won by pouring a solution of potassium hexacyanoferrate(II) over a Dower-50-exchanger and then poured into the 2.0 M HCl/TiCl<sub>4</sub> solution. The filling material from the exchanger is centrifused out after 24 h and dried over phosphorous pentaoxide and potassium hydroxide in a vacuum desiccators. The dried product was washed with water free from chloride ions and then dried again in the vacuum desiccators. The dried product was ground and sieved to 125 µm BSS mesh size.

# 2.3 Characterization of metal ferrocyanides

Manganese, silver ferrocyanides are found to have light blue colour, while titanium ferrocyanide have forest green colour. All are amorphous solid and shows no X-ray pattern. The metal ferrocyanides were characterized on the basis of elemental and spectral studies. The percentage composition of metals were determined by IL - 751 atomic absorption spectrophotometer. Carbon, hydrogen and nitrogen analysis were carried out by CEST - 118, CHN analyzer. Percentage composition of all three metal ferrocyanides are given in Table 1.

Infrared spectra of the metal ferrocyanides were recovered in KBr disc on Beckman IR -20 spectrophotometer. All three metal ferrocyanides show a broad peak at  $3800~\rm cm^{-1}$  is characteristics of water molecules and OH groups. Also a peak at around  $1600~\rm cm^{-1}$  is due to H-O-H bending. A sharp band at  $2000~\rm cm^{-1}$ and a broad peak at  $600~\rm cm^{-1}$ were observed in all three metal ferrocyanides are characteristics of cyanide and Fe - C stretching, respectively. A band around  $500~\rm cm^{-1}$  is observed in all three metal ferrocyanides may be due to polymerization of metal - nitrogen bond.

### 2.4 Stability of metal ferrocyanides:

Commented [HK2]:  $60^{\circ}$ C

Commented [HK3]: spelling

All three metal ferrocyanides were found to be stable in acids (HCl, H2SO4, HNO3,

CH<sub>3</sub>COOH) bases (NaOH, KOH, NH<sub>4</sub>OH) in concentration range 0.5 – 2.0 M at room and

boiling temperature. Metal ferrocyanides unaffected by salt (LiCl, NaCl, KCl, NH<sub>4</sub>Cl, RbCl, CsCl, BaCl<sub>2</sub> and CaCl<sub>2</sub>) solutions at room temperature in concentration range of 0.5-2.0~M.

Metal ferrocyanides are also found to be stable in tap and Atlantic ocean water at room and boiling temperature. The change in colour of metal ferrocyanides of various conditions are may be due to loss of water molecules from the compound.

## 2.5 Preparation of natural antifungal extracts

The extraction from cassia obtusifolia was done by wet method. The leaves of the plant were picked and soaked by covering with 95% ethanol solvent for 24 h. The ethanol – extract was filtered using glass wool. The filtrate was vaporized using a rotovapourizer at 45 C until all the ethanol is removed. The final extract was considered as the stock solutions from which further dilutions would be made for analysis cassia alata, cassia obtusifolia, ocimum sanctum and azadirachta indica extraction were done using dry method. The green leaves were dried for three days at 45 C, then grounded using an electric mill. The powdered leaves were then be soaked by covering with 95% ethanol solvent for 24 h. The soaking was repeated three times. The ethanol – extracts were then filtered by gravitation filtration using whatman filter paper. The ethanol was then be removed using rotovaporizer until the extract solidifies. The antifungal activity of plant extracts, metal ferrocyanides and plant extract – metal ferrocyanide complexes, were tested on a known cultured fungus, Aspergillus niger.

#### 2.6 Test on antifungal activity

2.6.1 Testing the antifungal activity of metal ferrocyanides only Metal ferrocyanide (10 mg) was placed in a sterilized petri dish containing media.

The fungal spores were then sprayed on the entire bottom of the dish using an aspirator. The similar method was repeated using different metal ferrocyanides.

2.6.2 Testing the antifungal activity of extract only

The antifungal plant extract (10 mg) was placed by means of washing with 20 ml ethanol in a sterilized petri dish containing media. The fungal spores were sprayed on the entire bottom of petri dish using an aspirator. The same method was repeated using different plants extract.

- 2.6.3 Testing the antifungal activity of antifungal plant extract metal ferrocyanide complexes Metal ferrocyanide (10 mg) and antifungal plant extract (10 mg) were placed in sterilized petri dish containing media. The fungal spores were sprayed on the entire bottom of the petri dish using an aspirator. This method was repeated using different extract and metal ferrocyanide complexes.
- 2.6.4 Testing the antifungal activity of control (ethanol only) Ethanol (20 ml) was placed in sterilized petri dish containing media. The fungal spores were then sprayed on the entire bottom of the dish using an aspirator. The essay was left to stand in sealed container in an incubator at 28 C for 168 h.

# 3. RESULTS AND DISCUSSION

3.1 Antifungal activity of metal ferrocyanides only

Antifungal activities of manganese, silver and titanium ferrocyanides was studied. Titanium ferrocyanide and manganese ferrocyanide were found to have maximum and minimum antifungal property respectively. The following order of antifungal activity was observed in metal ferrocyanides:

Titanium ferrocyanide > silver ferrocyanide > manganese ferrocyanide

The observation of bioassay test of metal ferrocyanides with fungal spores are given in Table 3.

# 3.2 Antifungal activity of extract only

Antifungal activity of azadirachta indica, cassia alata, cassia obtusifolia, ocimum sanctum and targetes were studied. Azadirachta indica and cassia obtusifolia were found to have maximum

and minimum antifungal property respectively. The following order of antifungal property was

Commented [HK4]: Which media?

Commented [HK5]: assay?

Commented [HK6]: 28°C

Azadirchta indica > tagetes patula > oscimum sanctum > cassia alata > cassia obtusifolia

The observations of bio assay test of natural antifungal extract only with cultured fungal spore are given in Table 4.

## 3.3 Antifungal activity of metal ferrocyanides and metal antifungal complexes The following order of antifungal activity was observed in natural antifungal with metal ferrocyanide complexes.

Manganese ferrocyanide
 Azadirachta indica > tagetes patula > ocimum sanctum > cassia alata > cassia obtusifolia

(ii) Silver ferrocyanide
Azadirachta indica > tagetes patula > ocimum sanctum > cassia alata > cassia obtusifolia

(iii) Titanium ferrocyanide

Azadirachta indica > tagetes patula > ocimum sanctum > cassia alata > cassia obtusifolia

Titanium ferrocyanide – azadirachta indica and manganese ferrocyanide – cassia obtusifolia complexes were found to have maximum and minimum antifungal properties, respectively. The observation of bioassay test of metal ferrocyanide – natural antifungal complexes with cultured fungal spore are given in Table 5.

# 3.4 Antifungal activity of control (ethanol only)

It was observed that fungal spores were able to grow in the control. The growth of fungal spore was unaffected by ethanol.

Table 1. Elemental analysis of manganese, silver and titanium ferrocyanides

Metal	Percentage found				
Ferrocyanides*	Metal	Iron	Carbon	Hydrogen	Nitrogen
MnFc	26.90	13.12	16.30	2.80	18.60
AgFc	40.22	8.75	11.47	2.75	13.77
TiFc	25.35	11.95	15.62	3.17	18.25

\*MnFc = Manganese ferrocyanide; AgFc = Silver ferrocyanide;

TiFc = Titanium ferrocyanide

Table 2. Infrared spectral data of manganese, silver and titanium ferrocyanides

Metal	Absorption frequencies (cm <sup>-1</sup> )				
Metai	H <sub>2</sub> O molecules/	НОН			
Ferrocyanides			$r C \equiv N$	↑ Fe – C	Metal – N*
	OH groups	bending			
MnFc	3800	1600	2000	610	500
AgFc	3800	1600	2010	600	490
TiFc	3800	1615	2020	600	500

<sup>\*</sup> metal – nitrogen band due to polymerization

Table 3. Observation of bioassay test of metal ferrocyanides only with cultured fungal spores

Manganese ferrocyanide	Silver ferrocyanide	Titanium ferrocyanide	
Evidence of few fungal spores growth was seen	Few spores of fungus were seen in area where silver ferrocyanide was not present.	No evidence of fungal spores growth seen.	

Bio assay: 10 mg metal ferrocyanide per petri dish

Room temperature:  $30 \pm 1 \ C$ 

Time: 168 h

Cultured fungus: Aspergillus niger

Order of antifungal activity: TiFc > AgFc > MnFc

Table 4. Observations of bioassay of natural antifungal extract only with cultured fungal spore

Azadirachta indica	Cassia alata	Cassia obtusifolia	Ocimum sanctum	Tagetes patula
Evidence of small	Evidence of more	Evidences of wide	Evidences of fungal	Evidences of small
amount of spores was	fungal spores was	spread fungal spores	spores was seen but	amount of spores
seen. It had least	seen but less than	was seen in		was seen but more
evidence of fungal	cassia obtusifolia	comparison to al	more than tagetes	than azadirachta
spores.	extract	other plant extract.	patula plant extract.	indica plant extract.

Bio assay: 10 mg plant extract per petri dish

Room temperature:  $30 \pm 1 \text{ C}$ 

Time: 168 h

Cultured fungus: Aspergillus niger

Order of antifungal activity: Azadirachta indica > targets patula > oscimum sanctum > cassia alata

> cassia obtusifolia

Table 5. Observations of bioassay test of metal ferrocyanides and natural anti fungal extract with cultured fungal spore

MFc	Azadirachta indica	Cassia alata	Cassia obtusifolia	Ocimum sanctum	Tagetes patula
MnFc	Little sign of fungal growth seen	Some fungal spores seen growing but more than O. Sanctum	Clumps of fungus were seen in the petri dish maximum growth	Few fungal spores seen growing more than T. Patula	Evidence of small amount of fungal growth but more than A. Indica.
AgFc	Very little sign of fungal growth seen	More spores were seen growing than O. Sanctum	Some spores were seen growing in clumps.	Evidences of small amount of fungal growth but more than T. Patula	Evidences of small amount of fungal growth but more than A. Indica
TiFc	No evidence of fungal growth maximum inhibition	Evidences of small amount of fungal growth but more than O. Sanctum	Evidences of small amount of fungal growth but more than C. Alata	Few fungal spores seen growing but more than T. Patula	Little fungal spores seen growing

Bio assay: 10 mg metal ferrocyanide plus 10 mg antifungal plant extract per petri dish

Room temperature:  $30 \pm 1$  C

Time: 168 h

Cultured fungus: Aspergillus niger

## CONCLUDING REMARKS

The following conclusions can be drawn from the present studies

- (a) Antifungal activity of secondary metabolities are enhanced through interaction with metal ferrocyanides.
- (b) Titanium and manganese ferrocyanides were found to have maximum and minimum antifungal property, respectively.
- (c) Azadirachta indica and cassia obtusifolia were found to have maximum and minimum antifungal properties, respectively.
- (d) Azadirachta indica extract titanium ferrocyanide complex and cassia obtusifolia extract manganese ferrocyanide complex were found to have maximum and minimum antifungal property respectively.
- (e) It may be also concluded from present studies that titanium ferrocyanide azadiracha indica extract complex may be used as effective medicine for skin infection.

# REFERENCE

- 1. P. Holliday, Fungal diseases of Tropical Crops, Cambridge University Press, UK, 1980.
- S. Palanichamy and S. Nagarajan, Analgesic activity of Cassia Alata Leaf Extract and Kaempferol 3-0-sopharoside, *J. Ethanopharmacol*, Elsevier, Ireland, 29 (1) (1990) 73-76
- 3. S. Palanichamy, Wound Healing Activity of Cassia Alata, Fitoterapia 62 (2) (1991) 336-338.

- H.M.H. Wijayakusuma, Medical plants in Indonesia, Part-I, Jakarta, Pustaka Kartini, 1996.
- S. K. Shaki, A. C. Shukla, A. K. Bajaj, U. Banerjee, D. Rimek, G. Midgely, A. Dikshit, Broad Spectrum herbal theraphy against superficial fungal infections, *Skin pharmacol. Appl. Skin Physiol*. 13 (2000) 60-64.
- J. B. Bohr, Chitin Synthase as an antifungal target: Recent Advances Curr. Med. Chem. Anti Infective Agents 2 (2) (2003) 173 – 789
- A. R. Jalilian, D. Sardari, S. Saradari, Application of Radioisotopes in antifungal research and fungal diseases studies, Curr. Med. Chem. Anti – Infective Agents 3 (4) (2004) 325 – 338.
- 8. H. J. Schoonbeek, J. M. Raaijaakers, M. A. De Waard, Fungal ABC transporters and microbial interactions in natural environments, *MPME* 15 (2002) 1165 1172
- S. Palamichamy, Effects of cassia alata leave extract on hyperglycemic rats, *J. Ethanopharmacol*. Elsevier, Ireland 29 (1990) 73 –78.
- C. J. Alexopoulos, C. W. Mims, M. Blackwell, *Characteristics of Fungi*, 4<sup>th</sup> ed., John Wiley and Sons Inc., New York, (1996) 30 – 803
- 11. G. Smith, An Introduction to Industrial Mycology, 6thed., London, 1969, 161.
- B. B. Tewari and N. Hamid, Interaction of glycine and β-alanine with nickel, cobalt and cadmium ferrocyanides, Colloids surf: Physiochem. Eng. Aspects 296 (2007) 264.
- 13. W. U. Malik, S. K. Srivastava, U. M. Bhandari, S. Kumar, Ion-exchange properties of chromium ferrocyanide, *J. Inorg. Nucl. Chem.* 38 (1976) 342 343.
- B. B. Tewari and Kamaluddin, Photosensitized oxidation of diphenylamine using nickel ferrocyanide and its relevance in chemical evolution, *Proceedings of Ninth National Space Science Symposium (NSSS – 96)*, Osmania University, Hyderabad, India, 1996, p. 93.
- V. Kourim, J. Rais and B. Million, Exchange properties of complex cyanides I (Ion exchange of caesium on ferrocyanides), J. Inorg. Nucl. Chem. 26 (1964) 1111.
- J. Bastian and K. H. Lieser, Ion Exchange properties of an titanium hexacyanoferrate(II), J. Inorg. Nucl. Chem. 29 (1967) 827 – 832.

1/24/2009